

Studies on the Radical Cyclization of 3-Oxopropanenitriles and Alkenes with Cerium(IV) Ammonium Nitrate in Ether Solvents

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The radical cyclization of 3-oxopropanenitriles **1a–1e** and alkenes **2a–2g** with cerium(IV) ammonium nitrate (CAN) in ether solvents was investigated (*Tables 1* and *2*). In the optimization study, 1,3-dioxolane, 1,4-dioxane, 1,2-dimethoxyethane, Et₂O, and THF were used as ether-based solvents, and the latter was found to be the most effective solvent in radical cyclizations mediated by cerium(IV). This system (cerium(IV)/THF) was applied to cyclizations of various 3-oxopropanenitriles and 1,3-dicarbonyl compounds with alkenes resulting in the formation of 4,5-dihydrofurans in high yields (*Table 2* and *Scheme 2*). The results of the cerium(IV)/THF radical cyclization were compared with those obtained with manganese(III) acetate/AcOH; the cerium(IV)/THF system turned out to be much more efficient.

Introduction. – During the last three decades, manganese(III) acetate dihydrate (MAH) [1] and cerium(IV) ammonium nitrate (CAN) [2] have been used as efficient radical oxidants. These oxidants have the ability for a one-electron transfer and are used in organic synthesis for constructing C–C bonds. Both CAN and MAH are well known for dihydrofuran formation by radical cyclization of 1,3-dicarbonyl compounds with alkenes. Dihydrofurans show a wide range of biological activities and form the basic structure of many natural compounds [3].

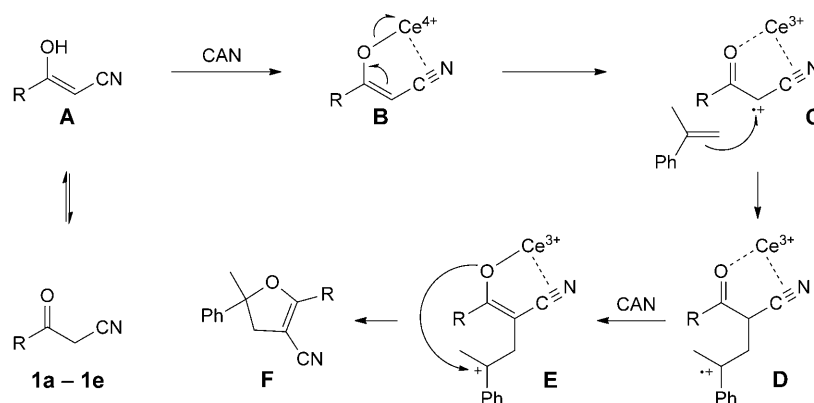
Our research group has studied the radical cyclization of 1,3-dicarbonyl compounds and 3-oxopropanenitriles with alkenes, alkynes, unsaturated amides, and dienes in the presence of manganese(III) acetate [4]. Very recently, we have reported that some 4,5-dihydrofuran-3-carbonitriles show antibacterial and antifungal activity [5]. These carbonitriles are available by oxidative cyclization of 3-oxopropanenitriles with conjugated alkenes [4f][4g] and unsaturated amides [4h]. In these reactions, MAH/AcOH gives good results, whereas CAN is not efficient when MeOH or MeCN are used as solvents. Hence, we performed an optimization using ether-based solvents and CAN which were more effective. We applied the optimized conditions (CAN/THF, 40–60°) to a study of the cyclization of 3-oxopropanenitriles and 1,3-dicarbonyl compounds with alkenes. Additionally, we compared cyclizations induced by CAN/THF and MAH/AcOH. The CAN/THF method gave 4,5-dihydrofurans in high yields.

Results and Discussion. – All 3-oxopropanenitriles were prepared according to [4g][4h][6]. The 1,1-diphenylbut-1-ene (**2e**) [7], 1,1-bis(4-fluorophenyl)ethene (**2f**) [8], and 1,1-bis(4-methylphenyl)ethene (**2g**) [9] were synthesized by dehydration of the corresponding alcohols obtained from *Grignard* reactions of suitable arylmagnesium bromides and ketones. The other alkenes, CAN, and reagents are commercial materials

and were purchased in highest purity. Crude products were purified by column chromatography or/and prep. thin-layer chromatography (TLC). All novel compounds were characterized by IR, ^1H - and ^{13}C -NMR, MS, and microanalysis. The other products were characterized by ^1H -NMR.

The proposed mechanism for the radical cyclization of 3-oxopropanenitriles with alkenes mediated by CAN and the formation of 4,5-dihydrofuran-3-carbonitriles is displayed in *Scheme 1*. According to this mechanism, $[\text{Ce}^{\text{IV}}(\text{enolato})]$ complex **B** is formed with the enol form **A** of 3-oxopropanenitriles **1a–1e** and CAN. While Ce^{4+} is reduced to Ce^{3+} , a radical at $\text{C}(\alpha)$ is formed resulting in structure **C**. The radical intermediate **D** is obtained by an electrophilic radical addition of **C** on the alkene. A $[\text{Ce}^{3+}(\text{enolato})]$ complex is formed by removing $\text{H}-\text{C}(\alpha)$ from structure **D**, and the radical is oxidized to carbocation **E** with CAN. The intramolecular cyclization of **E** forms 4,5-dihydrofuran-3-carbonitrile **F**.

Scheme 1. Mechanism for the Formation of 4,5-Dihydrofuran-3-carbonitriles



The synthesis of 4,5-dihydro-2,5,5-triphenylfuran-3-carbonitrile (**3a**) was used as model reaction. This compound **3a** was previously obtained by cyclization of 3-phenyl-3-oxopropanenitrile (**1a**) with 1,1-diphenylethene (**2a**) mediated by MAH in 65% yield [10]. For this model reaction, 1,3-dioxolane, 1,4-dioxane, 1,2-dimethoxyethane, Et_2O , and THF were used as ether solvents as well as MeOH and MeCN, which are traditionally used in reactions of CAN. The syntheses of **3a** were performed with a molar ratio **1a/2a/CAN** of 1.2:1:3, under N_2 at different temperatures for 10 min. The results are given in *Table 1*.

The formation of **3a** with CAN in MeOH at 0° and 60° failed, and compound **3a** could not be isolated. Dihydrofuran **3a** was formed in low yields in MeCN under the same reaction conditions. However, when 1,3-dioxolane and 1,4-dioxane were used at $40\text{--}50^\circ$, **3a** was obtained in 52 and 83% yield, respectively. At higher temperatures, lower yields were observed. In 1,2-dimethoxyethane at $40\text{--}50^\circ$, the yield of **3a** amounted to 73%. On the other hand, product **3a** was obtained in 84% yield in Et_2O under reflux for 24 h instead of 10 min (due to the poor solubility of CAN). The optimum yield of 97% was obtained in THF at $40\text{--}50^\circ$ after 10 min reaction time. These results indicated that ether solvents can be used more efficiently in reactions of

Table 1. Solvent- and Temperature-Optimization Study of the Radical Cyclization

$\text{Ph-C(=O)-CH}_2\text{-CN} + \text{Ph-CH=CH}_2 \xrightarrow{\text{CAN}} \text{Ph-C(Ph)(O-CH}_2\text{-C(=CN)-CH}_2\text{)-CH}_2\text{-Ph}$

1a **2a** **3a**

Solvent	T [°]	Yield [%] ^{a)}	Solvent	T [°]	Yield [%] ^{a)}
MeOH	0	–	1,3-dioxolane	60–70	48
MeOH	60	–	1,4-dioxane	40–50	83
MeCN	0	23	1,4-dioxane	60–70	52
MeCN	60	46	1,2-dimethoxyethane	40–50	73
1,3-Dioxolane	40–50	52	Et ₂ O	36	84
1,3-Dioxolane	50–60	50	THF	40–50	97

^{a)} Yield of isolated product based on alkene **2a**.

CAN. These optimum conditions were applied to the reactions of various 3-oxopropanenitriles, and the results were compared to the reactions performed with MAH/AcOH (Table 2).

While the radical cyclization of 3-(furan-2-yl)-3-oxopropanenitrile (**1b**) mediated by MAH gave **3b** in 50% yield, the new method (CAN/THF) produced **3b** in 90% yield (Table 2, Entry 2). Similarly, radical cyclizations of 3-(1-benzofuran-2-yl)-3-oxopropanenitrile (**1c**) and 4,4-dimethyl-3-oxopentanenitrile (**1d**) with **2a** in the presence of CAN gave the corresponding dihydrofurans in high yield (Entries 3 and 4). Reactions of **1a** with styrene (**2b**) and (1*E*)-1-phenylprop-1-ene (**2c**) which are less reactive than **2a**, furnished **3e** (42%) and **3f** (75%) in moderate to good yield (Entries 5 and 6). Whereas the reaction of 2-phenylprop-1-ene (**2d**) with **1a** and MAH/AcOH produced **3h** in 25% yield, the same reaction with CAN resulted in 80% yield (Entry 8). The reactions of the 3-oxopropanenitriles **1b**, **1c**, and **1e** with **2d** gave remarkable results (Entries 9–11). These facts evidently demonstrate that CAN/THF is a method superior to MAH/AcOH. Radical cyclizations of **1a** and **1e** with sterically hindered alkene **2e** showed again that CAN/THF is more efficient (Table 2, Entries 12 and 13). Moreover, 3-oxopropanenitriles **1a** and **1d** and alkenes **2f** and **2g** with 4-F and 4-Me substituents, respectively, gave 4,5-dihydrofuran-3-carbonitriles in similarly high yields (Entries 14 and 19).

This method could also be employed for other 1,3-diketones. The reaction of 4-hydroxycoumarin (**4**) and **2d** with CAN/THF at 40–50° for 10 min gave 2-methyl-2-phenyl-4*H*-furo[3,2-*c*][1]benzopyran-4-one (**4a**) in 72% yield (Scheme 2). Ketone **4a** has been synthesized in lower yields in the presence of Ag/*Celite* (37% [11] and 46% [12]) as well as in the presence of CAN under different reaction conditions (62% [2c] and 70% [2k]). Moreover, the radical cyclization of ethyl 3-oxobutanoate (**5**) with **2d** under the same conditions (CAN/THF) gave ethyl 4,5-dihydro-2,5-dimethyl-5-phenylfuran-3-carboxylate (**5a**) in 79% yield. This compound has been obtained in the ionic liquid [bmim][BF₄]/CH₂Cl₂ 1:9 as a solvent in the same yield but after 2 h (bmim = 1-butyl-9-methyl-1*H*-imidazolium) [2k]. A similar radical cyclization with 1,1-diphenyl-

Table 2. Radical Cyclization of 3-Oxopropanenitriles and Alkenes with CAN/THF

$$\text{R}^1\text{-C(=O)-CH}_2\text{-CN} + \text{R}^2\text{-C(R}^3\text{)=C(R}^4\text{)} \xrightarrow[60^\circ]{\text{CAN, THF}} \text{R}^1\text{-C(R}^2\text{)=C(R}^3\text{)-C(R}^4\text{)-CN}$$

1a – 1e **2a – 2g** **3a – 3t**

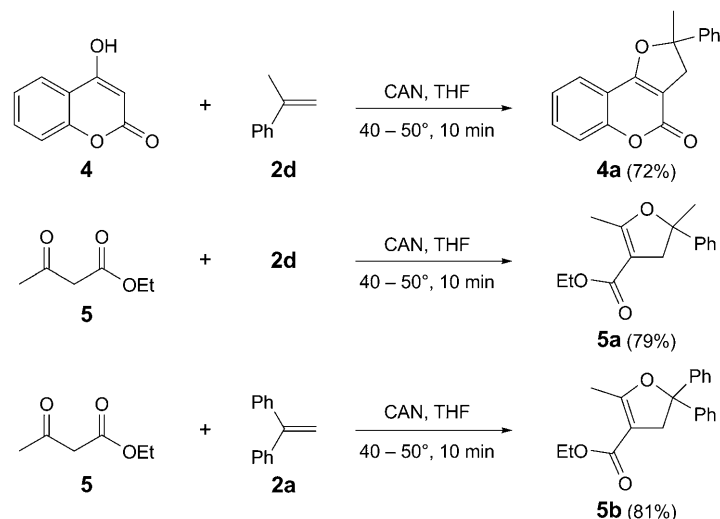
Entry	Nitrile	R ¹	Alkene	R ²	R ³	R ⁴	Product	Yield [%] ^{a)}	Yield [%] ^{b)} MAH/AcOH
1	1a	Ph	2a	Ph	Ph	H	3a	97	65
2	1b	furan-2-yl	2a				3b	90	50
3	1c	1-benzofuran-2-yl	2a				3c	96	
4	1d	^t Bu	2a				3d	90	62
5	1a	Ph	2b	Ph	H	H	3e	42	12
6	1a	Ph	2c	Ph	H	Me	3f	75	
7	1e	2-thienyl	2c				3g	77	
8	1a	Ph	2d	Ph	Me	H	3h	80	25
9	1b	furan-2-yl	2d				3i	77	
10	1c	1-benzofuran-2-yl	2d				3j	81	
11	1e	2-thienyl	2d				3k	86	
12	1a	Ph	2e	Ph	Ph	Et	3l	83	63
13	1e	2-thienyl	2e				3m	80	44
14	1a	Ph	2f	4-FC ₆ H ₄	4-FC ₆ H ₄	H	3n	86	62
15	1b	furan-2-yl	2f				3o	82	
16	1c	1-benzofuran-2-yl	2f				3p	88	
17	1e	2-thienyl	2f				3q	85	
18	1b	furan-2-yl	2g	4-MeC ₆ H ₄	4-MeC ₆ H ₄	H	3r	82	
19	1d	^t Bu	2g				3s	80	
20	1e	2-thienyl	2g				3t	87	

^{a)} Yields of isolated product based on the alkene. ^{b)} Yields taken from [10][4f].

ethylene (**2a**) produced **5b** in 81% yield, which has been obtained in 75% yield in ionic liquid [2k] and in 44% yield in MeCN by CAN [13].

Conclusions. – In this study, CAN was applied to the radical cyclizations of 3-oxopropanenitriles **1a–1e** with alkenes **2a–2g** for the first time. MeOH and MeCN, generally used for the reactions of CAN, either were not sufficiently effective or failed to form the products. An optimization study with ether solvents established that THF, Et₂O, and 1,4-dioxane are highly efficient in the reactions of CAN. Furthermore, 4,5-dihydrofuran-3-carbonitriles were obtained in higher yields with CAN than with MAH. The high efficiency of CAN/THF in radical cyclizations can be explained by the low solubility of CAN in THF that causes a stepwise release of Ce⁴⁺ ions into the reaction medium and thus prevents the formation of possible by-products.

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Scheme 2. Radical Cyclization of **4** and **5** with Alkenes

Experimental Part

General. All solvents were dried by standard methods. Column chromatography (CC): silica gel (SiO_2 ; 230–400 mesh; Merck). TLC: anal. aluminium plates coated with SiO_2 60 $F_{254-366}$ (0.2 mm; Merck). M.p.: electrothermal melting point apparatus; uncorrected. IR Spectra: FT-IR Matson-1000 instrument; KBr pellets; 400–4000 cm^{-1} range with 4 cm^{-1} resolution; $\tilde{\nu}$ in cm^{-1} . ^1H - and ^{13}C -NMR Spectra: Mercury-400 high-performance digital FT-NMR instrument; in CDCl_3 ; δ in ppm rel. to Me_4Si as internal standard, J in Hz. LC/ESI-MS: Waters-2695-Alliance-Micromass-ZQ instrument; in m/z (rel. %).

4,5-Dihydrofuran-3-carbonitriles 3a–3f. General Procedure. To a soln. of 3-oxopropanenitrile (1.2 mmol) and alkene (1 mmol) in THF (10 ml) under N_2 , a mixture of CAN (3 mmol) and NaHCO_3 (3 mmol) was added at 40° . Then, the temp. was slowly increased to 60° . The reaction was completed when the orange color of CAN had disappeared (10–30 min) or when the alkene spot on TLC had completely vanished. H_2O was added to the soln., and the mixture was extracted with CHCl_3 ($3 \times 20 \text{ ml}$). The combined org. phase was dried (Na_2SO_4) and concentrated and the crude product purified by CC (SiO_2 , 230–400 mesh) or prep. TLC ($20 \times 20 \text{ cm}$ plates, 2 mm thickness, hexane/AcOEt 5:1).

4,5-Dihydro-2,5,5-triphenylfuran-3-carbonitrile (3a): Yield 0.313 g (97%). Colorless solid. M.p. $134-136^\circ$ (hexane/AcOEt). ^1H -NMR: 8.10 (*dd*, $J = 7.0, 2.7, 2$ arom. H); 7.50–7.28 (*m*, 13 arom. H); 3.78 (*s*, $\text{CH}_2(4)$).

2-(Furan-2-yl)-4,5-dihydro-5,5-diphenylfuran-3-carbonitrile (3b): Yield 0.282 g (90%). Pale yellow oil. ^1H -NMR: 7.63 (*d*, $J = 1.6, 1$ H); 7.40–7.30 (*m*, 12 arom. H); 7.11 (*d*, $J = 3.5, 1$ H); 6.56 (*dd*, $J = 3.5, 1.6, 1$ H); 3.73 (*s*, $\text{CH}_2(4)$).

2-(Benzofuran-2-yl)-4,5-dihydro-5,5-diphenylfuran-3-carbonitrile (3c): Yield 0.348 g (96%). Yellow solid. M.p. $165-167^\circ$ (hexane/AcOEt). IR: 3061, 2200 (CN), 1648 (C=C), 1180 (C–O–C), 829, 750, 698. ^1H -NMR: 7.67 (*d*, $J = 7.8, 1$ H); 7.62 (*d*, $J = 8.2, 1$ H); 7.47–7.44 (*m*, 6 arom. H); 7.40–7.35 (*m*, 4 arom. H); 7.38–7.28 (*m*, 3 arom. H); 3.79 (*s*, $\text{CH}_2(4)$). ^{13}C -NMR: 157.0 (C(2)); 155.7; 144.9; 143.5; 128.8; 128.5; 127.5; 127.2; 125.9; 124.1; 122.5; 121.1; 116.2 (CN); 112.3; 110.3; 94.5; 81.2 (C(5)); 45.5 (C(4)). LC/ESI-MS: 364 (100, $[M+H]^+$). Anal. calc. for $\text{C}_{25}\text{H}_{17}\text{NO}_2$: C 82.63, H 4.72, N 3.85; found: C 82.86, H 4.37, N 4.05.

2-(tert-Butyl)-4,5-dihydro-5,5-diphenylfuran-3-carbonitrile (3d): Yield 0.273 g (90%). Yellow oil. ^1H -NMR: 7.40–7.48 (*m*, 10 arom. H); 3.58 (*s*, $\text{CH}_2(4)$); 1.36 (*s*, ^tBu).

4,5-Dihydro-2,5-diphenylfuran-3-carbonitrile (3e): Yield 0.103 g (42%). Yellow oil. ¹H-NMR: 8.02 (*dd*, *J* = 8.0, 5.0, 2 arom. H); 7.48–7.36 (*m*, 8 arom. H); 5.83 (*dd*, *J* = 11.2, 8.4, H–C(5)); 3.53 (*dd*, *J* = 14.8, 10.8, H_a–C(4)); 3.10 (*dd*, *J* = 14.8, 8.4, H_b–C(4)).

4,5-Dihydro-4-methyl-2,5-diphenylfuran-3-carbonitrile (3f): Yield 0.195 g (75%). Yellow oil. IR: 3030, 2207 (CN), 1646 (C=C), 1232 (C–O–C), 758, 706. ¹H-NMR: 8.02 (*dd*, *J* = 8.0, 2.0, 2 arom. H); 7.50–7.36 (*m*, 8 arom. H); 5.25 (*d*, *J* = 8.8, H–C(5)); 3.37 (*dq*, *J* = 8.8, 6.8, H–C(4)); 1.44 (*d*, *J* = 6.8, Me). ¹³C-NMR: 166.0 ((C(2))); 139.7; 131.7; 129.1; 129.0; 128.9; 128.1; 127.4; 125.9; 117.4 (CN); 91.8; 85.7 (C(5)); 47.9 (C(4)); 18.8 (Me). LC/ESI-MS: 262 (100, [M + H]⁺). Anal. calc. for C₁₈H₁₅NO: C 82.73, H 5.79, N 5.36; found: C 82.42, H 5.76, N 5.53.

4,5-Dihydro-4-methyl-5-phenyl-2-(2-thienyl)furan-3-carbonitrile (3g): Yield 0.205 g (77%). Yellow oil. IR: 3026, 2207 (CN), 1650 (C=C), 1244 (C–O–C), 758, 702. ¹H-NMR: 7.90 (*dd*, *J* = 3.6, 0.8, 1 arom. H); 7.52 (*dd*, *J* = 5.2, 1.2, 1 arom. H); 7.42–7.36 (*m*, 5 arom. H); 7.15 (*dd*, *J* = 5.2, 4.4, 1 arom. H); 5.26 (*d*, *J* = 8.4, H–C(5)); 3.36 (*dq*, *J* = 8.4, 6.4, H–C(4)); 1.43 (*d*, *J* = 6.4, Me). ¹³C-NMR: 161.4 (C(2)); 139.4; 130.3; 130.2; 130.1; 129.15; 129.1; 128.3; 126.0; 117.1 (CN); 92.5; 84.1 (C(5)); 47.7 (C(4)); 18.8 (Me). LC/ESI-MS: 268 (100, [M + H]⁺). Anal. calc. for C₁₆H₁₃NOS: C 71.88, H 4.90, N 5.24, S 11.99; found: C 71.74, H 4.68, N 5.40, S 12.32.

4,5-Dihydro-5-methyl-2,5-diphenylfuran-3-carbonitrile (3h): Yield 0.209 g (80%). Yellow oil. ¹H-NMR: 8.05 (*dd*, *J* = 7.81, 1.95, 2 arom. H); 7.49–7.31 (*m*, 8 arom. H); 3.31 (*d*, *J* = 14.5, H_a–C(4)); 3.23 (*d*, *J* = 14.5, H_b–C(4)); 1.40 (*s*, Me).

4,5-Dihydro-5-methyl-5-phenyl[2,2'-bifuran]-3-carbonitrile (3i): Yield 0.193 g (77%). Yellow oil. IR: 3061, 2209 (CN), 1648 (C=C), 1180 (C–O–C), 758, 702. ¹H-NMR: 7.61 (*d*, *J* = 1.2, 1 arom. H); 7.39–7.32 (*m*, 5 arom. H); 7.06 (*d*, *J* = 3.6, 1 arom. H); 6.65 (*dd*, *J* = 3.6, 1.6, 1 arom. H); 3.26 (*d*, *J* = 14.4, H_a–C(4)); 3.19 (*d*, *J* = 14.8, H_b–C(4)); 1.08 (*s*, Me). ¹³C-NMR: 156.9 (C(2)); 145.2; 144.7; 143.7; 128.6; 127.8; 124.1; 116.6 (CN); 113.9; 111.8; 91.1; 77.5 (C(5)); 45.2 (C(4)); 28.9 (Me). LC/ESI-MS: 252 (100, [M + H]⁺). Anal. calc. for C₁₆H₁₃NO₂: C 76.48, H 5.21, N 5.57; found: C 76.76, H 5.01, N 5.64.

2-(1-Benzofuran-2-yl)-4,5-dihydro-5-methyl-5-phenylfuran-3-carbonitrile (3j): Yield 0.244 g (81%). Pale yellow solid. M.p. 104–106° (hexane/AcOEt). IR: 3082, 2198 (CN), 1619 (C=C), 1211 (C–O–C), 856, 760, 696. ¹H-NMR: 7.66 (*d*, *J* = 7.6, 1 arom. H); 7.61 (*d*, *J* = 8.0, 1 arom. H); 7.43–7.37 (*m*, 5 arom. H); 7.35–7.28 (*m*, 3 arom. H); 3.32 (*d*, *J* = 15.2, H_a–C(4)); 3.25 (*d*, *J* = 14.8, H_b–C(4)); 1.85 (*s*, Me). ¹³C-NMR: 157.2 (C(2)); 155.6; 145.1; 144.8; 128.9; 128.1; 127.5; 127.1; 124.3; 124.0; 122.4; 116.5 (CN); 112.3; 110.2; 91.6; 80.7 (C(5)); 45.8 (C(4)); 29.1 (Me). LC/ESI-MS: 302 (100, [M + H]⁺). Anal. calc. for C₂₀H₁₅NO₂: C 79.72, H 5.02, N 4.65; found: C 79.84, H 4.88, N 4.89.

4,5-Dihydro-5-methyl-5-phenyl-2-(2-thienyl)furan-3-carbonitrile (3k): Yield 0.229 g (86%). Yellow oil. IR: 3026, 2970, 2201 (CN), 1635 (C=C), 1180 (C–O–C), 764, 702. ¹H-NMR: 7.09 (*dd*, *J* = 4.0, 0.8, 1 arom. H); 7.50 (*dd*, *J* = 5.2, 0.8, 1 arom. H); 7.38–7.36 (*m*, 4 arom. H); 7.30 (*t*, *J* = 3.2, 1 arom. H); 7.14 (*dd*, *J* = 4.8, 4.0, 1 arom. H); 3.26 (*d*, *J* = 14.4, H_a–C(4)); 3.19 (*d*, *J* = 14.8, H_b–C(4)); 1.79 (*s*, Me). ¹³C-NMR: 161.2 (C(2)); 145.1; 130.6; 130.1; 129.8; 128.9; 128.3; 128.0; 124.3; 117.6 (CN); 91.1; 77.2 (C(5)); 45.8 (C(4)); 28.8 (Me). LC/ESI-MS: 268 (100, [M + H]⁺). Anal. calc. for C₁₆H₁₃NOS: C 71.88, H 4.90, N 5.24, S 11.99; found: C 71.61, H 4.76, N 5.17, S 12.13.

4-Ethyl-4,5-dihydro-2,5,5-triphenylfuran-3-carbonitrile (3l): Yield 0.291 g (83%). Pale yellow solid. M.p. 139–141° (hexane/AcOEt). ¹H-NMR: 8.26 (*dd*, *J* = 7.4, 1.7, 2 arom. H); 7.76 (*dd*, *J* = 7.5, 1.5, 2 arom. H); 7.68–7.66 (*m*, 3 arom. H); 7.56 (*tt*, *J* = 7.0, 1.6, 2 arom. H); 7.52–7.44 (*m*, 6 arom. H); 4.04 (*dd*, *J* = 7.7, 5.9, H–C(4)); 1.59–1.52 (*m*, MeCH₂); 1.14 (*t*, *J* = 7.4, MeCH₂).

4-Ethyl-4,5-dihydro-5,5-diphenyl-2-(2-thienyl)furan-3-carbonitrile (3m): Yield 0.257 g (72%). Yellow oil. ¹H-NMR: 7.88 (*d*, *J* = 3.8, 1 arom. H); 7.51 (*d*, *J* = 7.2, 1 arom. H); 7.48 (*d*, *J* = 5.0, 1 arom. H); 7.32 (*t*, *J* = 7.8, 1 arom. H); 7.26–7.19 (*m*, 7 arom. H); 7.09 (*t*, *J* = 4.8, 1 arom. H); 3.76 (*t*, *J* = 6.9, H–C(4)); 1.34–1.25 (*m*, MeCH₂); 0.89 (*t*, *J* = 7.4, MeCH₂).

5,5-Bis(4-fluorophenyl)-4,5-dihydro-2-phenylfuran-3-carbonitrile (3n): Yield 0.308 g (86%). Yellow oil. ¹H-NMR: 7.41–7.38 (*m*, 5 arom. H); 7.36 (*dd*, *J* = 9.0, 5.0, 4 arom. H); 7.12 (*t*, *J* = 8.4, 1 arom. H); 3.62 (*s*, CH₂(4)).

5,5-Bis(4-fluorophenyl)-4,5-dihydro[2,2'-bifuran]-3-carbonitrile (3o): Yield 0.286 g (82%). Yellow oil. IR: 3063, 2200 (CN), 1623 (C=C), 1253 (C–O–C), 829, 702. ¹H-NMR: 7.65 (*d*, *J* = 1.2, 1 arom. H); 7.40–7.36 (*dd*, *J* = 9.0, 5.0, 4 arom. H); 7.13 (*d*, *J* = 3.5, 1 arom. H); 7.08 (*t*, *J* = 8.6, 4 arom. H); 6.59 (*dd*,

$J = 3.5, 1.6, 1$ arom. H); 3.71 (s, CH₂(4)). ¹³C-NMR: 163.6–161.4 ($d, {}^1J(\text{C,F}) = 246.9, \text{C-F}$); 156.7 (C(2)); 145.8; 143.5; 139.2 ($d, {}^4J(\text{C,F}) = 3.2$); 127.8 ($d, {}^3J(\text{C,F}) = 8.3$); 116.2 (CN); 115.8 ($d, {}^2J(\text{C,F}) = 21.8$); 114.5; 112.2; 93.4; 78.1 (C(5)); 45.3 (C(4)). LC/ESI-MS: 350 (100, [M + H]⁺). Anal. calc. for C₂₁H₁₃F₂NO₂: C 72.20, H 3.75, N 4.01; found: C 72.44, H 3.52, N 4.30.

2-(1-Benzofuran-2-yl)-5,5-bis(4-fluorophenyl)-4,5-dihydrofuran-3-carbonitrile (3p): Yield 0.351 g (88%). Yellow solid. M.p. 104–106° (hexane/AcOEt). IR: 3059, 2200 (CN), 1643 (C=C), 1172 (C–O–C), 835, 754, 700. ¹H-NMR: 7.65 ($d, J = 7.6, 1$ arom. H); 7.59 ($dd, J = 8.4, 0.8, 1$ arom. H); 7.44 (s, 1 H); 7.41–7.36 ($m, 5$ arom. H); 7.29 ($t, J = 7.6, 1$ H); 7.06 ($t, J = 8.4, 4$ H); 3.73 (s, CH₂(4)). ¹³C-NMR: 163.9–161.4 ($d, {}^1J(\text{C,F}) = 246.9, \text{C-F}$); 156.8 (C(2)); 144.6; 139.1 ($d, {}^4J(\text{C,F}) = 3.1$); 127.8 ($d, {}^3J(\text{C,F}) = 7.7$); 127.4; 127.3; 124.1; 122.5; 115.8 ($d, {}^2J(\text{C,F}) = 21.3$); 115.9 (CN); 112.3; 110.6; 93.6; 81.1 (C(5)); 45.6 (C(4)). LC/ESI-MS: 400 (100, [M + H]⁺). Anal. calc. for C₂₅H₁₅F₂NO₂: C 75.18, H 3.79, N 3.51; found: C 75.52, H 3.66, N 3.42.

5,5-Bis(4-fluorophenyl)-4,5-dihydro-2-(2-thienyl)furan-3-carbonitrile (3q): Yield 0.310 g (85%). Yellow oil. IR: 3028, 2194 (CN), 1606 (C=C), 1257 (C–O–C), 835, 702. ¹H-NMR: 7.95 ($d, J = 3.9, 1$ H); 7.56 ($d, J = 4.7, 1$ H); 7.37 ($dd, J = 9.0, 5.08, 4$ H); 7.18 ($t, J = 3.9, 1$ H); 7.06 ($t, J = 8.6, 4$ H); 3.65 (s, CH₂(4)). ¹³C-NMR: 163.8–161.4 ($d, {}^1J(\text{C,F}) = 246.8, \text{C-F}$); 160.8 (C(2)); 139.3 ($d, {}^4J(\text{C,F}) = 3.2$); 130.4; 130.2; 130.0; 128.5; 127.7 ($d, {}^3J(\text{C,F}) = 8.4$); 116.9 (CN); 115.9–115.7 ($d, {}^2J(\text{C,F}) = 21.1$); 93.1; 77.6 (C(5)); 45.6 (C(4)). LC/ESI-MS: 366 (100, [M + H]⁺). Anal. calc. for C₂₁H₁₃F₂NOS: C 69.03, H 3.59, N 3.83, S 8.78; found: C 69.14, H 3.68, N 3.70, S 8.96.

4,5-Dihydro-5,5-bis(4-methylphenyl)[2,2'-bifuran]-3-carbonitrile (3r): Yield 0.280 g (82%). Yellow oil. IR: 3017, 2198 (CN), 1608 (C=C), 1255 (C–O–C), 835, 704. ¹H-NMR: 7.62 ($d, J = 1.5, 1$ H); 7.26 ($d, J = 7.8, 4$ H); 7.16 ($d, J = 7.6, 1$ H); 7.08 ($d, J = 3.5, 1$ H); 6.54 ($dd, J = 3.5, 1.56, 1$ H); 3.68 (s, CH₂(4)); 2.33 (s, 2 Me). ¹³C-NMR: 157.0 (C(2)); 145.6; 143.9; 138.1; 129.4; 125.8; 116.7 (CN); 114.2; 112.1; 94.3; 78.1 (C(5)); 45.1 (C(4)); 21.2 (Me). LC/ESI-MS: 342 (100, [M + H]⁺). Anal. calc. for C₂₃H₁₉NO₂: C 80.92, H 5.61, N 4.10; found: C 81.22, H 5.77, N 4.02.

2-(tert-Butyl)-4,5-dihydro-5,5-bis(4-methylphenyl)furan-3-carbonitrile (3s): Yield 0.265 g (80%). Yellow oil. IR: 3032, 2922, 2203 (CN), 1617 (C=C), 1111 (C–O–C), 827, 706. ¹H-NMR: 7.27 ($d, J = 8.0, 4$ H); 7.14 ($d, J = 7.6, 4$ H); 3.55 (s, CH₂(4)); 2.33 (s, 2 Me). ¹³C-NMR: 177.7 (C(2)); 141.8; 137.8; 129.3; 125.6; 117.5 (CN); 92.3; 77.4 (C(5)); 64.0; 35.0 (C(4)); 28.3; 21.2 (Me). LC/ESI-MS: 332 (100, [M + H]⁺). Anal. calc. for C₂₃H₂₅NO: C 83.34, H 7.60, N 4.23; found: C 83.54, H 7.46, N 4.36.

4,5-Dihydro-5,5-bis(4-methylphenyl)-2-(2-thienyl)furan-3-carbonitrile (3t): Yield 0.310 g (87%). Yellow oil. IR: 3078, 2205 (CN), 1618 (C=C), 1238 (C–O–C), 835, 696. ¹H-NMR: 7.92 ($dd, J = 4.0, 0.8, 1$ H); 7.52 ($dd, J = 4.4, 0.4, 1$ H); 7.27 ($d, J = 8.0, 4$ H); 7.15 ($d, J = 7.6, 4$ H); 3.70 (s, CH₂(4)); 2.32 (s, 3 Me). ¹³C-NMR: 160.8 (C(2)); 140.8; 137.9; 130.2; 129.9; 129.7; 129.2; 128.1; 125.5; 117.1 (CN); 93.8; 77.4 (C(5)); 45.3 (C(4)); 21.0 (Me). LC/ESI-MS: 358 (100, [M + H]⁺). Anal. calc. for C₂₃H₁₉NOS: C 77.28, H 5.36, N 3.92, S 8.97; found: C 77.38, H 5.11, N 4.11, S 9.12.

2,3-Dihydro-2-methyl-2-phenyl-4H-furo[3,2-c][1]benzofuran-4-one (4a): Yield 0.200 g (72%). Colorless solid. M.p. 114–116°. ¹H-NMR: 7.71 ($dd, J = 6.3, 1.5, 1$ H); 7.5 ($td, J = 8.65, 1.59, 1$ H); 7.37–7.26 ($m, 7$ arom. H); 3.35 ($d, J = 15.4, \text{H}_a\text{-C}(3)$); 3.26 ($d, J = 15.1, \text{H}_b\text{-C}(3)$); 1.83 (s, Me).

Ethyl 4,5-Dihydro-2,5-dimethyl-5-phenylfuran-3-carboxylate (5a): Yield 0.195 g (79%). Yellow oil. ¹H-NMR: 7.42 ($d, J = 8.0, 2$ H); 7.34 ($t, J = 7.6, 2$ H); 7.30 ($dd, J = 8.0, 1.2, 1$ H); 3.41 ($d, J = 15.0, \text{H}_a\text{-C}(4)$); 3.35 ($d, J = 15.0, \text{H}_b\text{-C}(4)$); 4.22 ($q, J = 7.2, \text{MeCH}_2$); 2.63 (s, Me); 1.34 ($t, J = 7.2, \text{MeCH}_2$).

Ethyl 4,5-Dihydro-2-methyl-5,5-diphenylfuran-3-carboxylate (5b): Yield 0.250 g (81%). Yellow oil. ¹H-NMR: 8.05 ($d, J = 7.6, 1$ H); 7.51–7.34 ($m, 8$ arom. H); 3.32 (s, CH₂(4)); 4.32 ($q, J = 7.6, \text{MeCH}_2$); 2.65 (s, Me); 1.35 ($t, J = 7.6, \text{MeCH}_2$).

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